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Hohmann transfer calculation to always find the first Extra-fixes for KSP 0.17 compatibility, Decoupled MechJeb from the SAS/ASAS key; Manual HDG/PTH entry;

http://wiki.mechjeb.com/index.php?title=Version_History

Piping Calculations Manual - Shashi Menon McGraw-Hill: Year: 01/01/2005: Pages: 666: Version: Steam System Piping Chapter 5:

<http://www.abe.pl/en/book/9780071440905/piping-calculations-manual>

The experiments in this manual have been in use in my introductory chemistry class calculate the mass of water in the unpopped Calculate the Ksp of calcium

<http://home.comcast.net/~mrstine/Chemistrylabmanual2.doc>

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McGraw-Hill. 2001 Kellogg Company. Design of piping systems. E.S. Piping calculations manual. McGraw-Hill. 2005 Nayar, M.L. Piping handbook. McGraw-Hill. 2000

<http://piping.unizar.es/libros-sobre-tuberias>

The equilibrium is represented by the equation above. (a) Write the expression for the solubility-product constant, Ksp, and calculate its value at 18 C. (b)

<http://chemistrygeek.com/study%20guides/AP%20Chemistry/AP%20Chapter%2014%20-%20Equilibrium%20and%20Ch%2016%20Ksp.doc>

Piping Calculations Manual by Shashi Menon, December 10, You could add Piping Calculations Manual (Mcgraw-Hill Calculations) to a list if you log in.

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Site offers a calculator to predict the effects of changing gearing, sprockets, drive ratio, tires, chain and maximum RPM.
<http://www.gearingcommander.com/>

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Olex's beautiful Interactive illustrated interplanetary guide and calculator which inspired me to create this tool as a web page.
<http://alexmoon.github.io/ksp/>

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The Ksp of cobalt(II) hydroxide, $\text{Co}(\text{OH})_2$, is 5.92×10^{-15} . Calculate the solubility of this compound in g/L. Answers. Get this Solutions Manual; Scholarships;

<http://www.chegg.com/homework-help/questions-and-answers/ksp-cobalt-ii-hydroxide-co-oh-2-592-x-10-15-calculate-solubility-compound-g-l-q5010290>

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<http://www.alibris.com/Piping-Calculations-Manual-E-Shashi-Menon/book/8287759>

1. Calculate the Ksp for CaCl_2 if 200.0g of CaCl_2 is required to saturate 100.0 mL of solution.

<http://krawchyk.sswiki.com/file/view/Solubility+Workbook.doc>

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<http://www.petrolcom.net/vb/threads/26542-Piping-and-Pipeline-Calculations-Manual-Construction-Design-Fabrication-and-Examina>

Calculate the molar solubility of Ag_2SO_4 ($K_{sp} = 1.4 \times 10^{-5}$) in: a.) Pure water; b.) 0.10 M Na_2SO_4 and c.) Solutions Manual; Scholarships; Career Search; Online Tutoring;

<http://www.chegg.com/homework-help/questions-and-answers/calculate-molar-solubility-agso4-ksp-14-10-5--pure-water-b-010-m-na2so4-c-010-m-agno3-q566742>

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